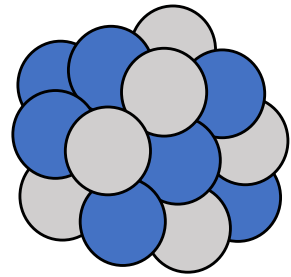
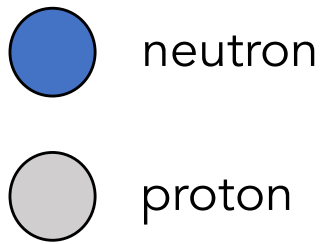




# Carbon Isotopes in Plants



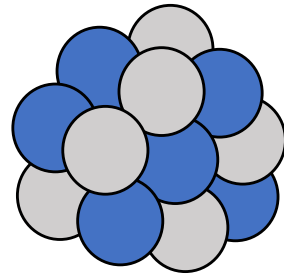




$^{13}\text{C}$

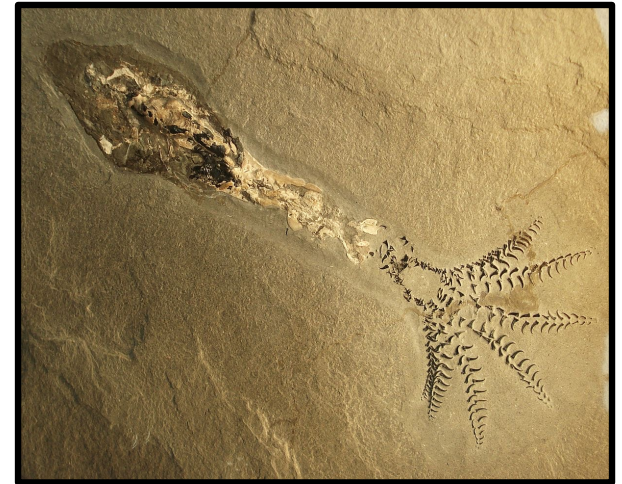
1.1% abundance

one extra neutron,  
8% heavier than  $^{12}\text{C}$ !



$^{12}\text{C}$

98.9% abundance



Vienna Bee Dee Belemnite

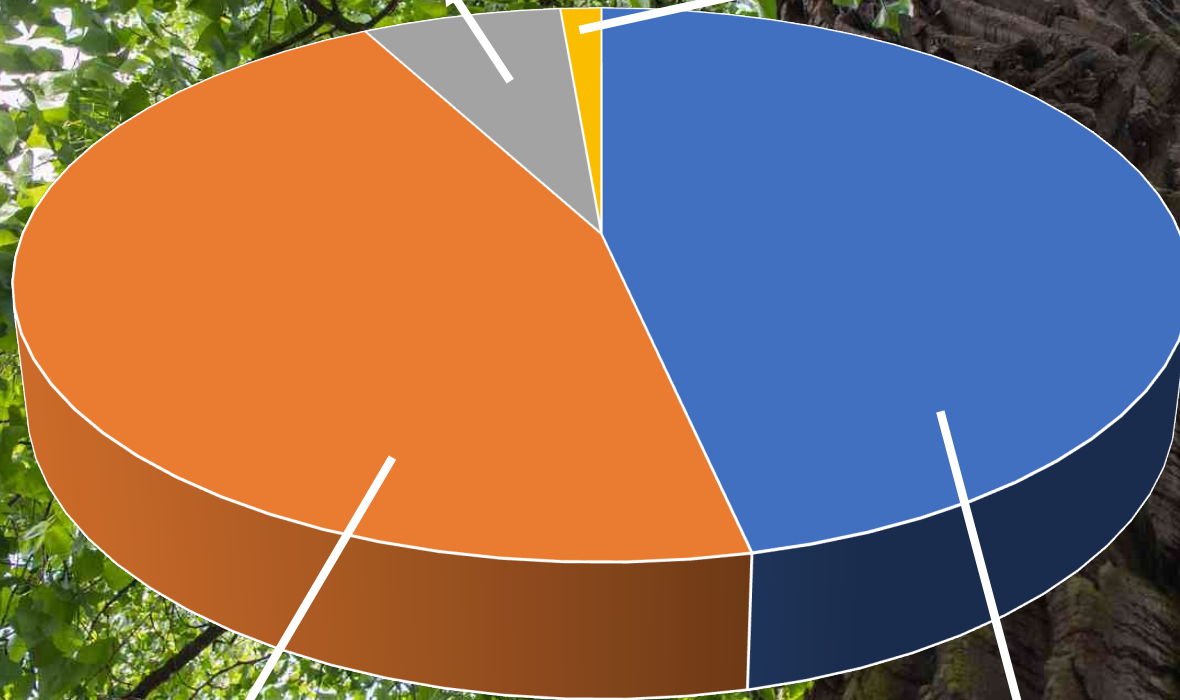
$$\delta^{13}\text{C} = \left( \frac{[^{13}\text{C}/^{12}\text{C}]_{\text{sample}}}{[^{13}\text{C}/^{12}\text{C}]_{\text{standard}}} - 1 \right) \times 1000$$

expressed as a permill (‰)

# What are plants made of?

**Hydrogen**  
6.3%

**Nitrogen**  
1.3%



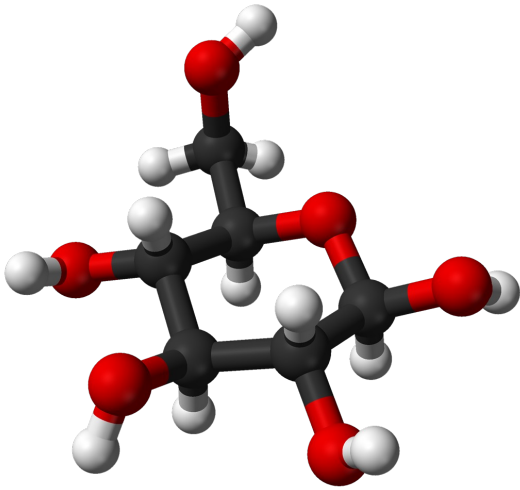
**Carbon**  
44%

**Oxygen**  
45%



# What are plants made of?

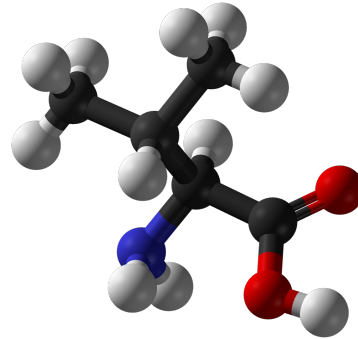
## Carbohydrates



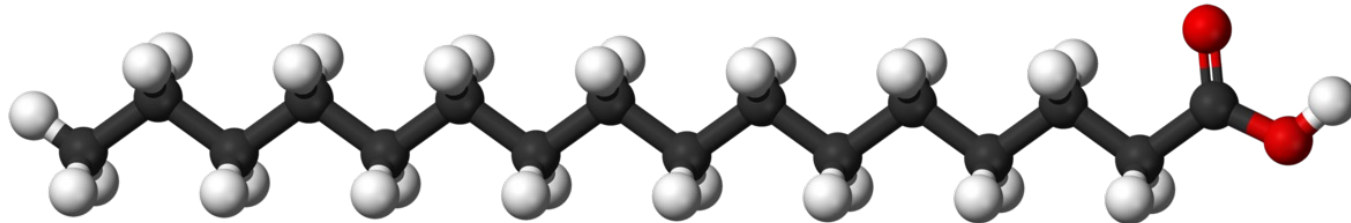
Glucose

Cellulose

## Proteins



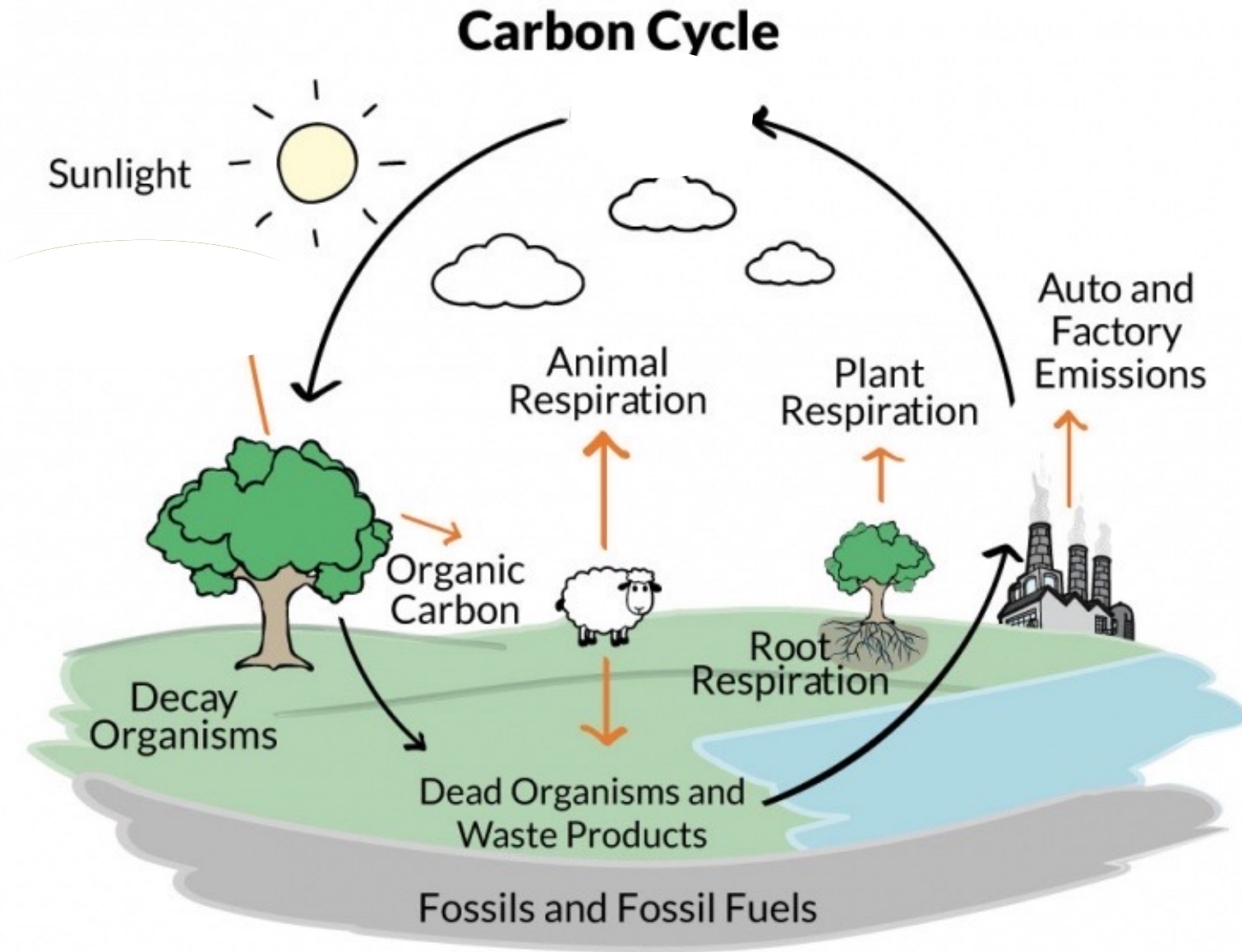
Amino Acids



Lipids



# How do plants get carbon?

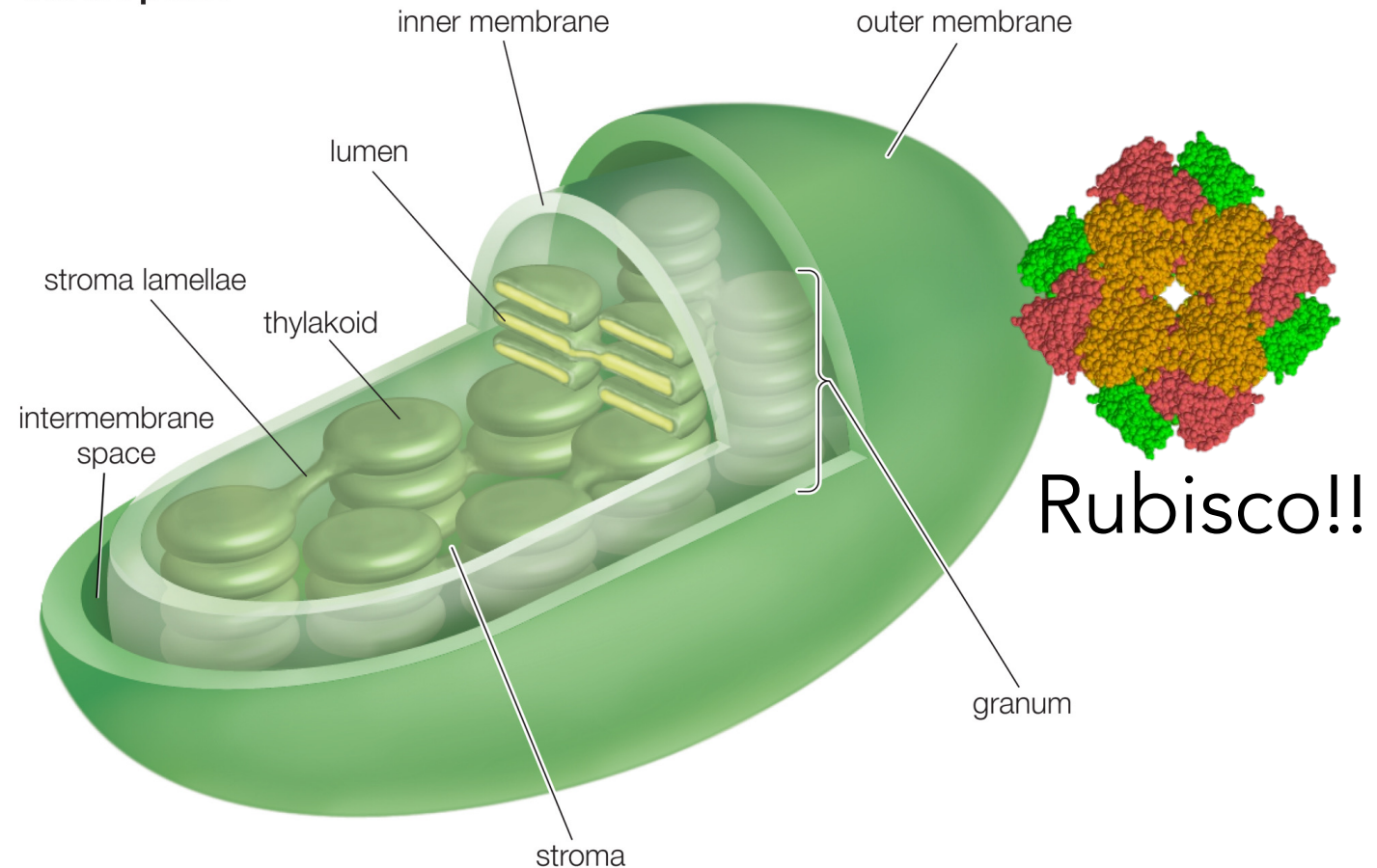




# Photosynthesis

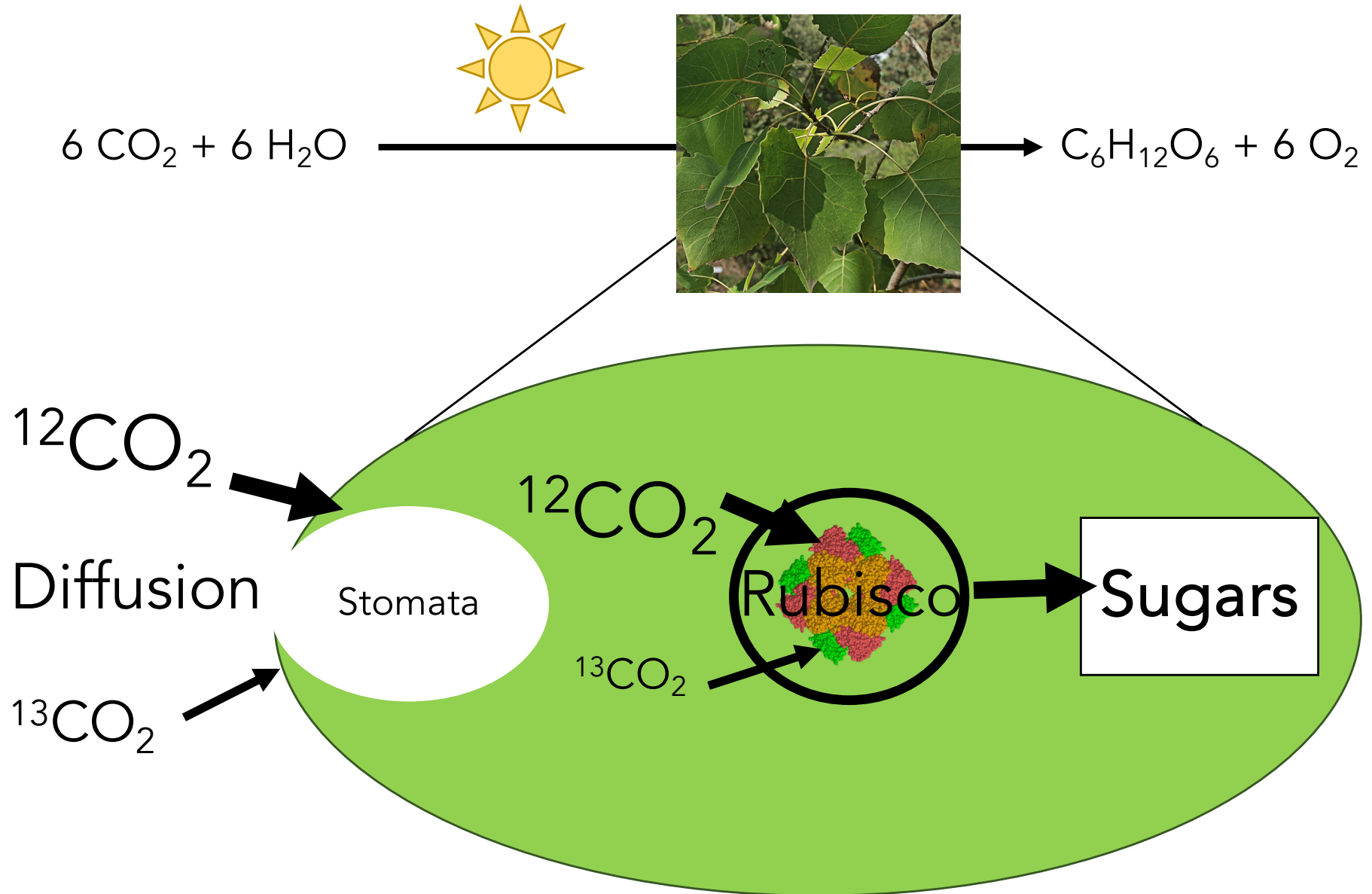


Chloroplast



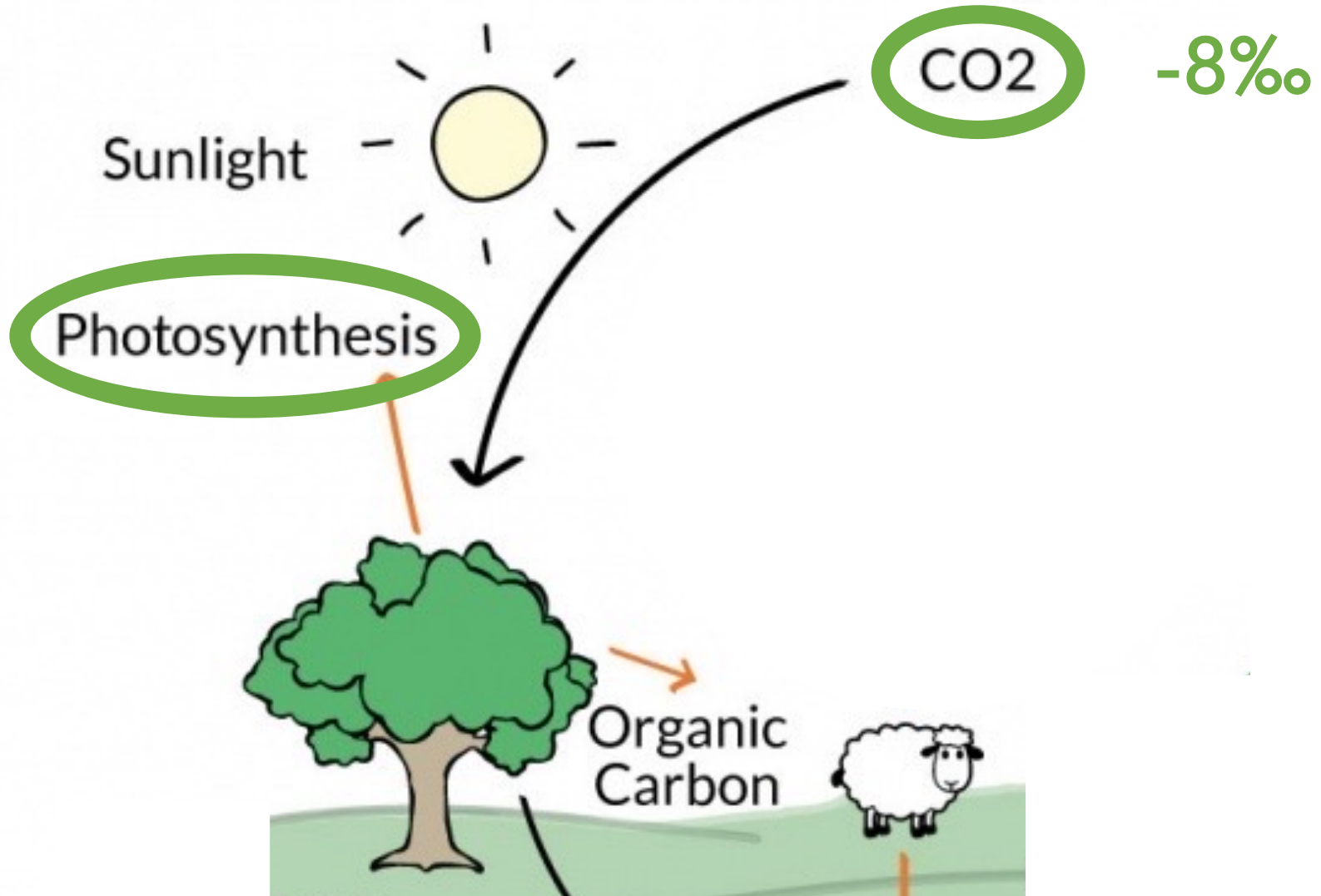


# Photosynthesis and Carbon Isotopes

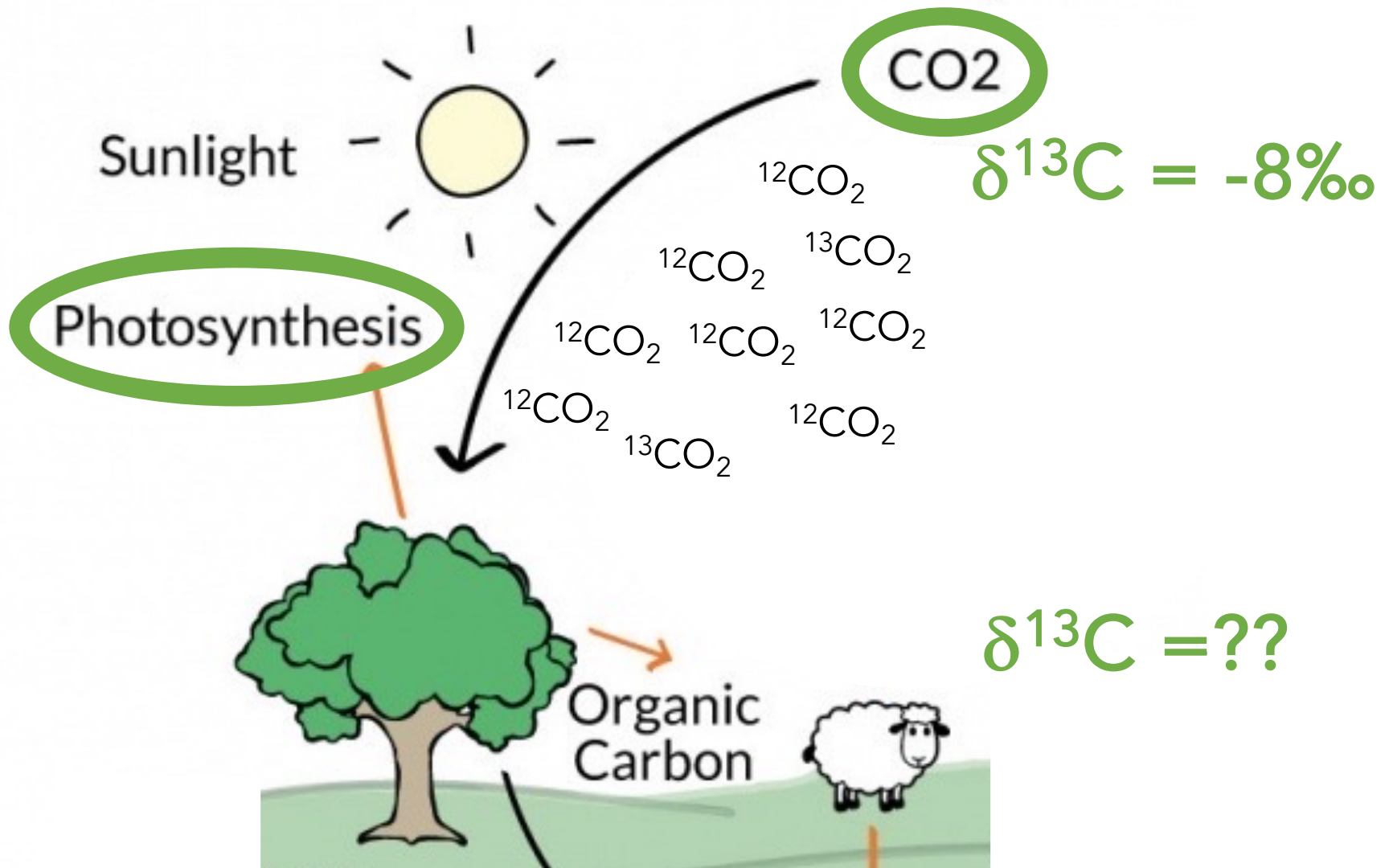




# Carbon Cycle



# Carbon Cycle







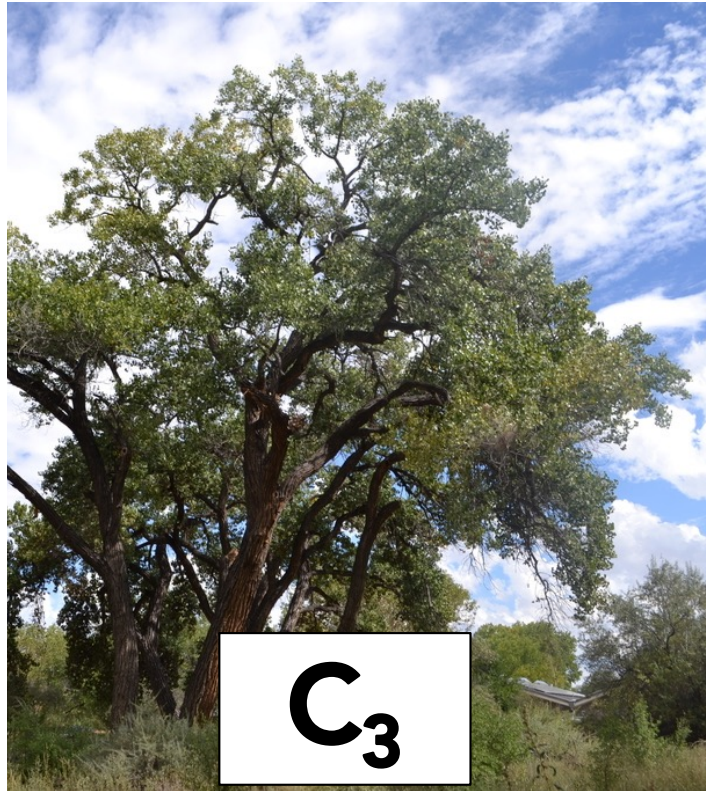
Plants have  
 $\delta^{13}\text{C}$  values  
around

-12 to -31‰

LOTS of  $^{12}\text{CO}_2$

Not much  $^{13}\text{CO}_2$

# $\delta^{13}\text{C}$ values can tell you about *how* plants are doing photosynthesis



**C<sub>3</sub>**

$\delta^{13}\text{C}$   
-30‰ to -20‰

Cottonwood



**CAM**

$\delta^{13}\text{C}$   
-27‰ to -12‰

Prickly Pear



**C<sub>4</sub>**

$\delta^{13}\text{C}$   
-15‰ to -9‰

Black Grama



# $\delta^{13}\text{C}$ values can tell you about *how* plants are doing photosynthesis

Rubisco has a high affinity for  $^{12}\text{CO}_2$

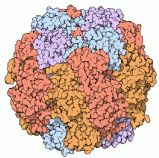
lower  $\delta^{13}\text{C}$  values

PEP carboxylase isn't as picky as Rubisco

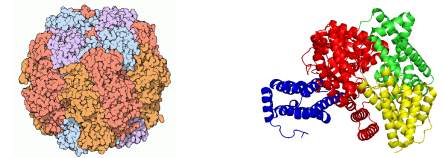
higher  $\delta^{13}\text{C}$  values



C

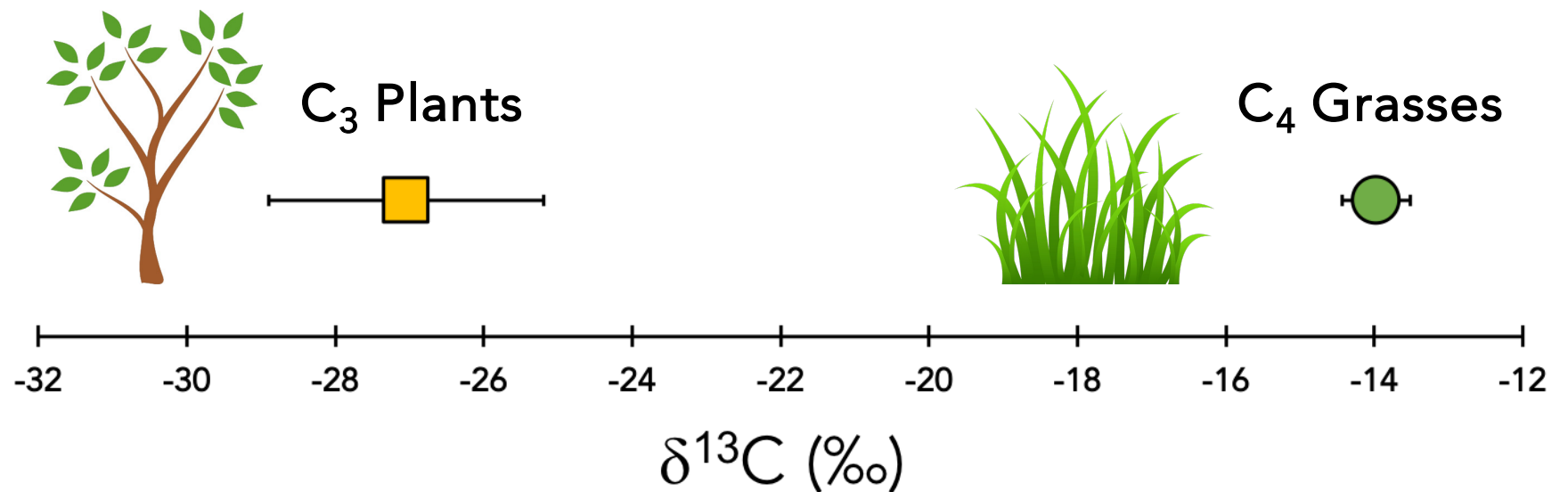


Rubisco



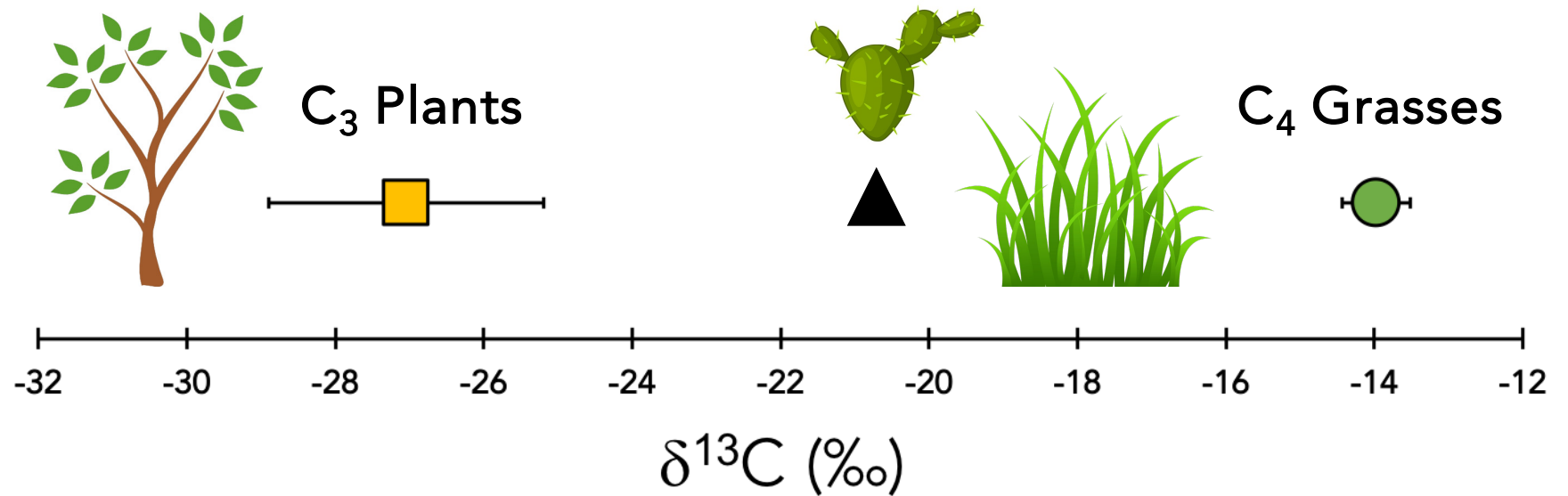
Rubisco & PEP  
Carboxylase

# Different types of plants have different $\delta^{13}\text{C}$ values

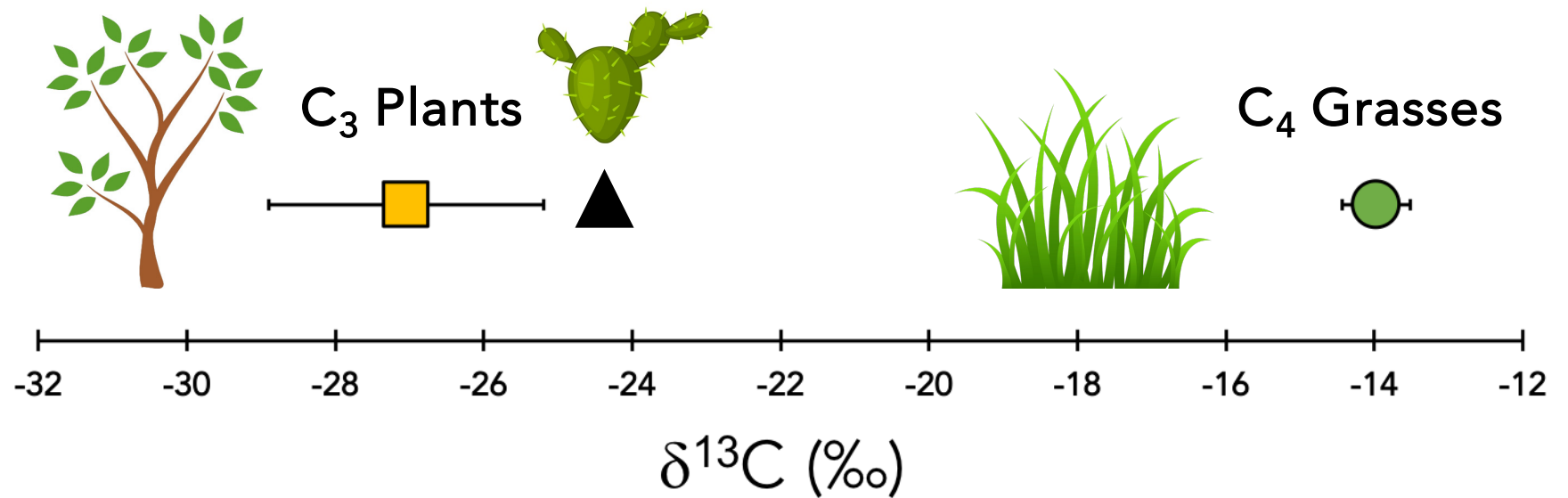




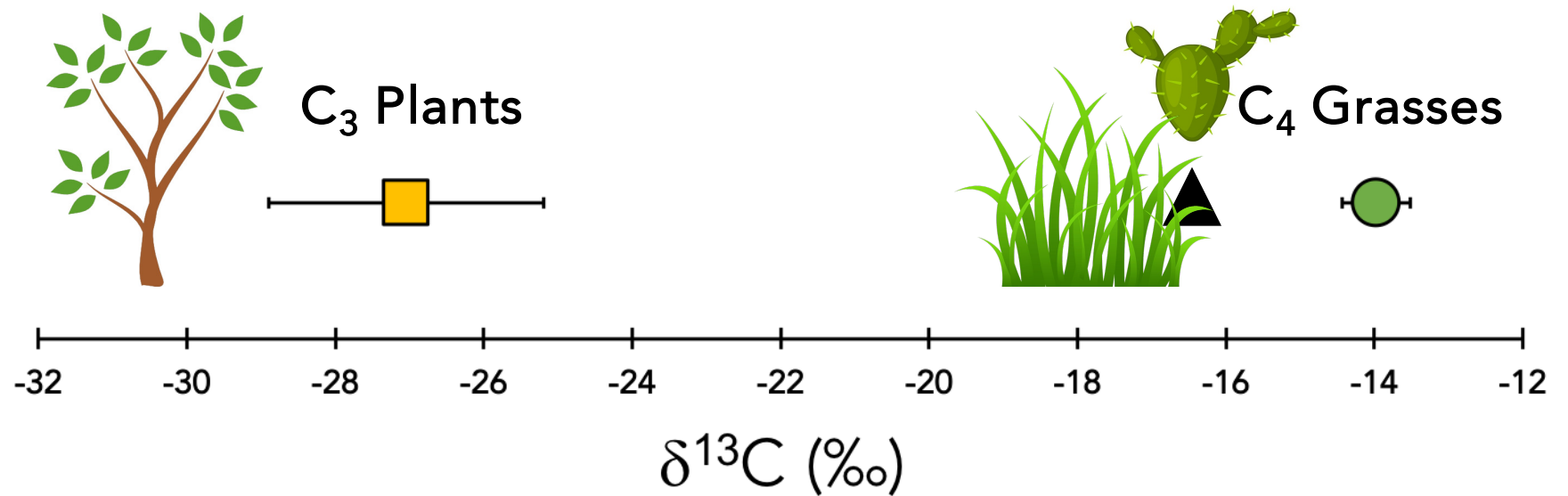
# Different types of plants have different $\delta^{13}\text{C}$ values



# Different types of plants have different $\delta^{13}\text{C}$ values

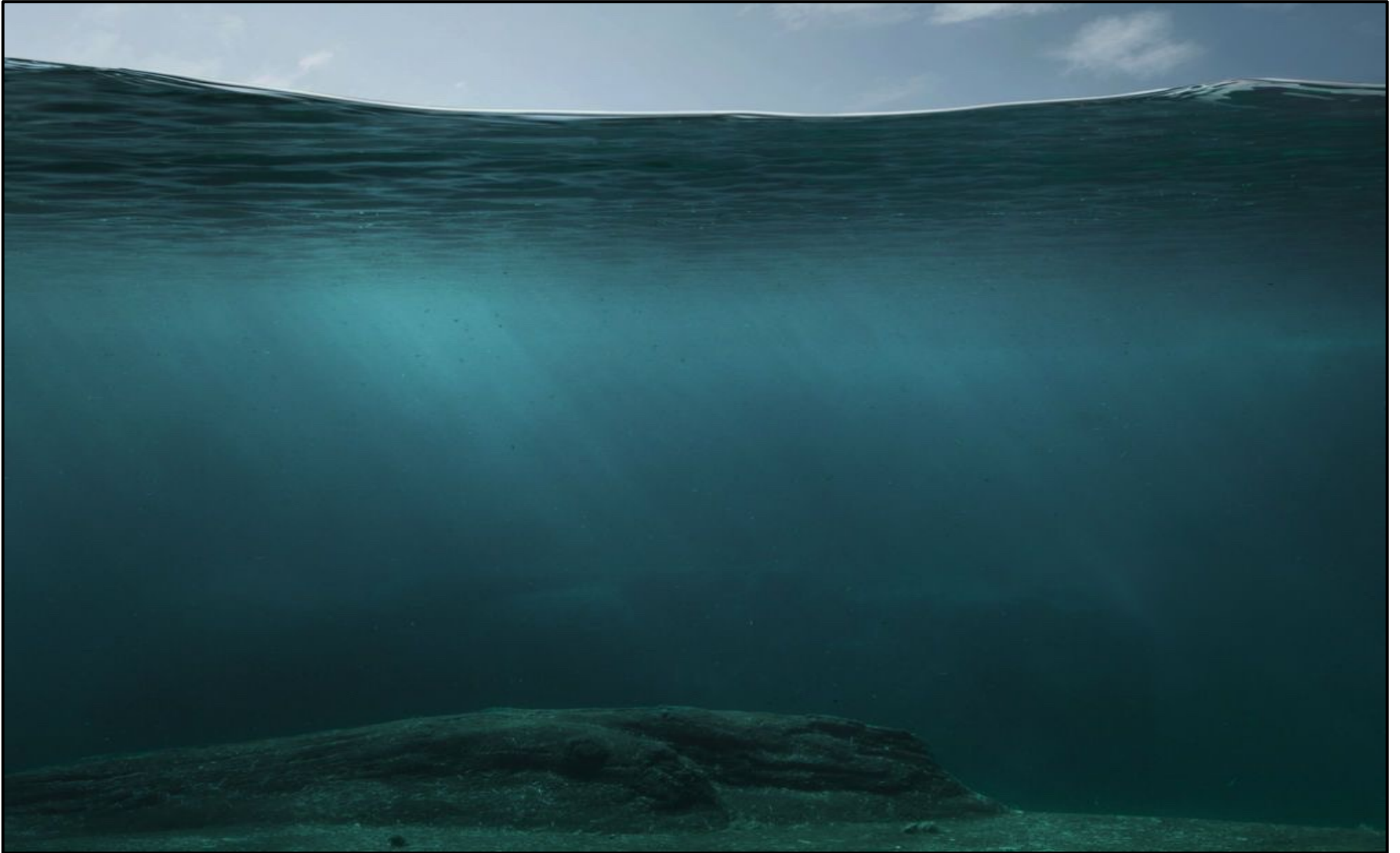


# Different types of plants have different $\delta^{13}\text{C}$ values





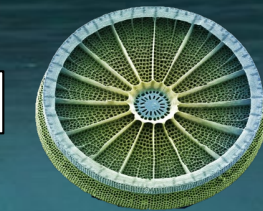
# Photosynthesis in water



# Aquatic Inorganic Carbon Sources

$\text{CO}_{2(\text{atm})} = -8\text{‰}$

$\text{CO}_{2(\text{diss})} = -9\text{‰}$



$\delta^{13}\text{C} \sim -24\text{‰}$

$\text{HCO}_3^- = 0\text{‰}$

$\text{CO}_3 = 1\text{‰}$

$\text{H}_2\text{CO}_3 = -7\text{‰}$



$\delta^{13}\text{C} \sim -16\text{‰}$





# TAKE HOME POINTS

- Plants build their tissues with carbon
- They do this via **PHOTOSYNTHESIS**
  - They take in  $\text{CO}_2$  and chemically convert it into sugars
- Because lighter goes faster, plants have more  $^{12}\text{C}$  than atmospheric  $\text{CO}_2$ 
  - So, they have  $\delta^{13}\text{C}$  values that are lower than the  $\delta^{13}\text{C}$  value of atmospheric  $\text{CO}_2$



# TAKE HOME POINTS

- The exact  $\delta^{13}\text{C}$  values of plants are determined primarily by the type of photosynthesis they do:  $\text{C}_3$ , CAM,  $\text{C}_4$
- For aquatic plants, the type of carbon they use impacts their  $\delta^{13}\text{C}$  values

